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Characterization of the MCR_{red2} form of methyl-coenzyme M reductase: a pulse EPR and ENDOR study

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Abstract Methyl-coenzyme M reductase (MCR), which catalyses the reduction of methyl-coenzyme M (CH_3 -S-CoM) with coenzyme B (H-S-CoB) to CH_4 and CoM-S-S-CoB, contains the nickel porphinoid F430 as prosthetic group. The active enzyme exhibits the Ni(I)-derived axial EPR signal MCR_{red1} both in the absence and presence of the substrates. When the enzyme is competitively inhibited by coenzyme M (HS-CoM) the MCR_{red1} signal is partially converted into the rhombic EPR signal MCR_{red2} . To obtain deeper insight into the geometric and electronic structure of the red2 form, pulse EPR and ENDOR spectroscopy at X- and Q-band microwave frequencies was used. Hyperfine interactions of the four pyrrole nitrogens were determined from ENDOR and HYSCORE data, which revealed two sets of nitrogens with hyperfine couplings differing by about

a factor of two. In addition, ENDOR data enabled observation of two nearly isotropic 1H hyperfine interactions. Both the nitrogen and proton data indicate that the substrate analogue coenzyme M is axially coordinated to Ni(I) in the MCR_{red2} state.

Electronic Supplementary Material Supplementary material is available for this article if you access the article at <http://dx.doi.org/10.1007/s00775-003-0450-y>. A link in the frame on the left on that page takes you directly to the supplementary material.

Keywords ENDOR · HYSCORE · Factor 430 · Methanogenic archaea · Nickel enzyme

Abbreviations CH_3 -S-CoM: methyl-coenzyme M · dq : double-quantum · ENDOR: electron nuclear double resonance · EPR: electron paramagnetic resonance · HS-CoB: coenzyme B · HS-CoM: coenzyme M · HYSCORE: hyperfine sublevel correlation · MCR: methyl-coenzyme M reductase · MCR_{red1} : MCR exhibiting the red1 EPR signal · MCR_{red2} : MCR exhibiting the red2 EPR signal · sq : single-quantum

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Introduction

Methyl-coenzyme M reductase (MCR) catalyzes the reduction of methyl-coenzyme M [CH_3 -S-CoM, 2-(methylthio)ethanesulfonate] with coenzyme B (HS-CoB, 7-thioheptanoylthreonine phosphate) to methane and CoM-S-S-CoB, which is the methane-forming step in the energy metabolism of all methanogenic archaea [1]. The enzyme contains the nickel porphinoid cofactor F430, which has been identified as the prosthetic group of MCR [2, 3, 4]. The crystal structure of inactive MCR reveals the presence of two symmetry-related active sites, each containing one cofactor F430 in an inactive and EPR-silent Ni(II) state (d^8 , $S=1$) [5, 6, 7, 8]. Different forms of the enzyme and interconversions between various MCR species are known [9, 10]. Several of them, including the states MCR_{ox1} , MCR_{ox2} ,

MCR_{red1}, and MCR_{red2}, and various subgroups of these forms, are EPR visible. The oxidation state of all the red forms is Ni(I) [11, 12, 13]. The active form, MCR_{red1}, exhibits an axial Ni-based EPR spectrum characteristic of a d^9 , $S=1/2$ Ni(I) complex with the unpaired electron in the $d_{x^2-y^2}$ orbital and hyperfine structure due to the four pyrrole nitrogen ligands. MCR_{red1} is converted into MCR_{red2} in the presence of coenzyme M (HS-CoM, 2-mercaptoethanesulfonate) and coenzyme B, the MCR_{red2} signal apparently being generated at the expense of the MCR_{red1} signal. At most, 50% of MCR_{red1} is found to be converted to MCR_{red2}. EPR data show clearly that MCR_{red2} is nickel based. First, the CW EPR spectrum of ^{61}Ni -labeled MCR_{red2} is considerably broadened due to a ^{61}Ni hyperfine interaction of $(A_1, A_2, A_3)=(39, 44, 67)$ MHz [9, 10]. Second, the g matrix is highly rhombic, indicating that the unpaired electron is in a low-symmetry environment. Third, the large deviation of the principal g values from the g value of the free electron is typical for a metal-based complex with a large spin-orbit coupling. In contrast to MCR_{red1} and MCR_{ox1}, the ligand hyperfine interactions are not resolved in the X-band or Q-band EPR spectra of MCR_{red2}. The poor resolution of the EPR spectra can be traced back to g and/or A strain caused by the environment surrounding the nickel ion not being as well defined as in the other EPR-active MCR species. It is thus necessary to use high-resolution techniques such as HSCORE (hyperfine sublevel correlation) and pulse ENDOR (electron nuclear double resonance) to obtain further information about the coordination sphere.

In this contribution we report on the magnetic parameters of the ligand nuclei of MCR_{red2}, in particular the hyperfine and nuclear quadrupole interactions of the “corphin” nitrogens and on proton hyperfine interactions, which have been determined to obtain a deeper insight into the geometric and electronic structure of this state of MCR. A combination of HSCORE and pulse ENDOR [14] experiments carried out at X- and Q-band frequencies enabled determination of the hyperfine interactions of the four pyrrole nitrogens, which differ in their hyperfine couplings by a factor of two. In addition, two large and nearly isotropic ^1H hyperfine couplings were observed. We tentatively assign these couplings to protons of the substrate analogue coenzyme M, which we assume to be axially bound.

Materials and methods

Sample preparation

The sample preparation is described elsewhere [9]. Note that a sample containing only MCR_{red2} cannot be prepared, since a significant amount of MCR_{red1} is always present. It is possible, however, to prepare a sample that only contains MCR_{red1}.

Spectroscopy

The X-band pulse ENDOR experiments were carried out on a Bruker ELEXSYS spectrometer [microwave (mw) frequency 9.73 GHz] equipped with a helium gas-flow cryostat (Oxford). All measurements were done at 15 K with a repetition rate of 200 Hz. Pulse measurements at Q-band (mw frequency 35.3 GHz) were performed on a home-built spectrometer [15] at 20 K with a repetition rate of 1 kHz.

Davies-ENDOR [16] spectra at X- and Q-band frequencies were measured with the mw pulse sequence π - T - $\pi/2$ - τ - π - τ -echo. At X-band, short mw pulses of length $t_{\pi/2}=26$ ns and $t_{\pi}=52$ ns were used to suppress signals from weakly coupled protons that overlap with the strongly coupled nitrogen signals. A time interval $\tau=900$ ns was used, and during time T an rf π pulse of length 5.4 μs with variable frequency ν_{ENDOR} and frequency increments of 50 kHz was applied. At Q-band, proton (nitrogen) spectra were measured with mw pulse lengths of $t_{\pi/2}=50$ (30) ns and $t_{\pi}=100$ (60) ns, and a time interval $\tau=220$ ns. An rf π pulse with a length of 18 (29) μs and frequency increments of 50 kHz was applied during time T .

HSCORE [17], with the pulse sequence $\pi/2$ - τ - $\pi/2$ - t_1 - π - t_2 - $\pi/2$ - τ -echo, is a two-dimensional experiment, which correlates nuclear frequencies in one electron spin manifold to nuclear frequencies in the other electron spin manifold. Measurements at Q-band were run using the following parameters: mw pulses of length $t_{\pi/2}=t_{\pi}=16$ ns, starting times of 96 ns for t_1 and t_2 , and time increments $\Delta t=8$ ns (data matrix 256×256). To increase the modulation depth, the second and the third $\pi/2$ pulses were replaced by a matched pulse of length 24 ns [18]. To avoid blind spots, spectra with τ values of 112, 132, 152, and 192 ns were recorded. An eight-step phase cycle was used to remove unwanted echoes.

Both ENDOR and HSCORE experiments were carried out at different magnetic field positions to collect data from different molecular orientations (orientation selectivity) [14].

Data manipulation

The data were processed with the program MATLAB 6.1 (MathWorks, Natick, Mass., USA). The time traces of the HSCORE spectra were baseline corrected with a third-order polynomial, apodized with a Hamming window and zero filled. After a two-dimensional Fourier transformation the absolute-value spectra were calculated. The HSCORE spectra recorded with different τ values were added to eliminate τ -dependent blind spots. The Davies-ENDOR spectra were simulated with the program EasySpin (<http://www.esr.ethz.ch>) and the HSCORE spectra with a program written in-house [19].

Theory

The spin Hamiltonian for a spin system with a Ni(I) ion (electronic configuration $3d^9$, $S=1/2$), ^{14}N ($I=1$), and ^1H ($I=1/2$) nuclei in frequency units is given by [14]:

$$H = (\beta_e/h)\mathbf{SgB}_0 + \sum \mathbf{SA}_i\mathbf{I}_i - (\beta_n/h)\sum g_{i,n}\mathbf{I}_i\mathbf{B}_0 + \sum \mathbf{I}_i\mathbf{Q}_i\mathbf{I}_i \quad (1)$$

where the terms describe the electron Zeeman interaction, the hyperfine interactions, the nuclear Zeeman interactions, and the nuclear quadrupole interactions (for nuclei with $I > 1/2$). The principal values Q_1 , Q_2 , and Q_3 of the traceless $\mathbf{Q}_i$ tensor are usually expressed in terms of the quadrupole coupling constant $K=e^2qQ/4I(2I-1)\hbar$ and the asymmetry parameter $\eta=(Q_1-Q_2)/Q_3$, with $Q_1=-K(1-\eta)$, $Q_2=-K(1+\eta)$, and $Q_3=2K$.

The ENDOR spectrum of a nucleus with spin $I=1/2$ consists of two transitions. For $\mathbf{B}_0$ along one of the hyperfine principal axes the frequencies are given by:

$$\nu = \left| \nu_I \pm \frac{A_K}{2} \right| \quad (2)$$

where ν_I is the nuclear Zeeman frequency and A_K is one of the principal hyperfine values. For $\nu_I > |A_K/2|$ (weak coupling case) the two lines are centered at ν_I and separated by A_K . In MCR_{red2} this is the case for the proton signals observed at X- and Q-band.

For a nucleus with spin $I=1$ (^{14}N) the ENDOR spectrum consists of four lines. For the magnetic field along one of the principal axes the frequencies are given by:

$$\nu = \left| \nu_I \pm \frac{A_K}{2} \pm \frac{3}{2} Q_K \right| \quad (3)$$

For frequencies along arbitrary field orientations, see [14]. For $\nu_I < |A_K/2|$ (strong coupling case) the lines are centered at $A_K/2$ and split by $2\nu_I$ and the nuclear quadrupole interaction.

In a HYSORE experiment on a nucleus with spin $I=1$ in each of the two electron spin manifolds (α, β) there exist three transitions between the nuclear sublevels, two single-quantum (sq) transitions with $|\Delta m_I|=1$, and one double-quantum (dq) transition with $|\Delta m_I|=2$. The two-dimensional HYSORE spectra are usually dominated by cross-peaks between the dq frequencies.

Results

The Q-band EPR spectra of MCR_{red1} and MCR_{red2} (and simulations) are shown in Fig. 1. The spectrum of MCR_{red1} overlaps with the spectrum of MCR_{red2}, except for the g_1 feature of MCR_{red2} at the low-field end. The spectrum of MCR_{red2} exhibits strong rhombicity with g values (g_1, g_2, g_3) = (2.2869, 2.2313, 2.1753). These values are more accurate than the X-band data [9] because of the higher resolution afforded by Q-band data (see Table 1).

Figure 2A and Fig. 2B show ENDOR spectra recorded at X-band (at a low-field position where only MCR_{red2} exists) and at Q-band (field positions indicated in Fig. 1), respectively. At X-band, signals from weakly coupled protons, which overlap with signals from the strongly coupled nitrogen(s), are suppressed by using short mw pulses (hyperfine contrast selective ENDOR [14]). The spectrum is thus dominated by signals from strongly coupled nitrogens and signals from two protons with a large hyperfine coupling (indicated by asterisks). In Fig. 2B (Q-band) the spectra arise only from nitrogens since the protons are shifted to higher frequencies (around 48 MHz; see Fig. 4). The low-field ENDOR spectrum (trace Ia) contains only MCR_{red2}, whereas traces IIa and IIIa contain contributions from both

Fig. 1 (a) First derivative of the Q-band FID-detected EPR spectrum measured at 20 K. (b) Simulation of MCR_{red2} and (c) MCR_{red1}. The labels *I* (1103.2 mT), *II* (1130.2 mT), and *III* (1155.7 mT, 1157.7 mT), indicate the field positions used in the ENDOR and HYSORE experiments, respectively

MCR_{red1} and MCR_{red2}. Traces IIc and IIIc were measured on a separate sample containing only the MCR_{red1} species, which allows the features originating from MCR_{red1} in traces IIa and IIIa to be identified. Simulations of both the X- and Q-band ENDOR spectra were achieved by considering three ^{14}N nuclei: two equivalent nitrogens with $A_{\text{iso}} = 24.6$ MHz and one nitrogen with $A_{\text{iso}} = 22.5$ MHz (see Table 2). At the low-field position the simulations of the individual components are given: Id ($\times 1$ nitrogen), Ic ($\times 2$ equivalent nitrogens), and Ib the sum. For trace Id a stick diagram is also given, showing how the nitrogen frequencies are determined by the hyperfine, nuclear Zeeman, and nuclear quadrupole interactions. In a powder sample, broad lines rather than sharp peaks are generally observed since many molecular orientations contribute to the spectrum at each field position; this makes interpretation without simulation difficult (in particular, trace IIa, measured along g_2). However, spectra recorded at the edges of the EPR spectrum (traces Ia and IIIa for MCR_{red2}) are “single-

Table 1 g values of MCR_{red2} in comparison with other MCR forms and Ni(I) complexes

Complex	g_1	g_2	g_3	Ref.
MCR _{red2} (Q-band)	2.2869(5)	2.2313(5)	2.1753(5)	This work
MCR _{red2} (X-band)	2.2880	2.2348	2.1790	[9]
MCR _{red1c}	2.250	2.071	2.0605	[9]
MCR _{red1}	2.2485	2.070	2.060	[20]
Ni(I)F430	2.244	2.063	2.063	[26]
MCR _{ox1}	2.2310	2.1667	2.1532	[9]
MCR _{ox1}	2.229	2.166	2.148	[20]
Ni(I)(STPP)(SO ₂)	2.075	2.087	2.187	[23]
Ni(I)(STPP)(2,4-Me ₂ py) ₂	2.262	2.233	2.131	[23]
[Ni(I)(DAPA)(SPh) ₂] [−]	2.283	2.201	2.164	[24]
(CTPP)Ni(III)OH	2.274	2.190	2.109	[25]

Fig. 2 A X-band Davies-ENDOR spectrum recorded at 303.7 mT: (a) experimental, (b) nitrogen simulation. **B** Q-band nitrogen Davies-ENDOR spectra recorded at the observer positions indicated in Fig. 1. Ia, IIa, and IIIa, experimental; Ib, Ic, Id, IIb, and IIIb, simulation; IIc and IIIc, experimental spectra of a sample containing only MCR_{red1}

crystal” like since they select only a narrow range of molecular orientations; the peaks here are thus sharper and amenable to a simple interpretation via a stick diagram. The complete set of ENDOR spectra and simulations are available as Supplementary material.

Q-band HYSCORE spectra measured at the observer positions indicated in Fig. 1 are shown in Fig. 3A. These spectra clearly reveal the presence of an additional nitrogen(s) with a smaller hyperfine coupling ($A_{\text{iso}} = 13.8$ MHz), which was barely observable in the ENDOR spectra. The nitrogens with the larger hyperfine couplings could also be observed in the HYSCORE spectra after matching the mw pulses to increase the

modulation depth. In Fig. 3A, selected cross-peaks from the two types of ^{14}N nuclei have been labeled. The peaks are predominantly found in the $(-, +)$ quadrant, indicating that for both types of nitrogens we have a strong-coupling situation, $A > 2\nu_I \approx 7$ MHz. The dq cross-peaks in particular are informative since along a principal axis they are centered at A , split by four times the ^{14}N nuclear Larmor frequency ($\nu_I \approx 3.5$ MHz) and free from nuclear quadrupole broadening (to first order). In Fig. 3AII the dq cross-peaks from the weaker coupled nitrogens are centered around $(-20.6, 7.0)$ MHz and $(-7.0, 20.6)$ MHz, and from the stronger coupled nitrogens around $(-31.2, 17.4)$ MHz and $(-17.4, 31.2)$ MHz. At the observer position in the middle of the EPR spectrum (Fig. 3AII), many molecular orientations contribute to the HYSCORE spectrum and thus ridges whose length reflects the anisotropy of the coupling parameters are observed. The HYSCORE spectra recorded at the edges of the EPR spectrum (Fig. 3AI and Fig. 3AIII for the MCR_{red2} species) are “single-crystal” like and the peaks are thus sharper. At field positions outside the EPR spectrum of MCR_{red2} the features from the nitrogen(s) with the smallest hyperfine coupling disappear, and only strongly coupled nitrogen signals remain consistent with the nitrogen couplings of MCR_{red1} given in [20]. Note that the diagonal peaks in the first quadrant are caused by an improper transfer of nuclear coherences between the two electron-spin manifolds by the π pulse and slight phase errors in the phase cycling procedure. In Fig. 3B, a simulation for observer position II (along g_2) is given which includes three stronger ($A_{\text{iso}} = 24.6, 24.6, 22.5$ MHz) and one weaker coupled ($A_{\text{iso}} = 13.8$ MHz) nitrogen (see Table 2). Simulations at the three field positions are given in the Supplementary material.

Figure 4 shows proton Davies-ENDOR spectra measured at Q-band; the observer positions are again those given in Fig. 1. Protons with nearly isotropic hyperfine couplings of $(A_1, A_2, A_3) = (10.1, 10.7, 10.3)$ MHz and $(A_1, A_2, A_3) = (6.5, 7.1, 6.6)$ MHz are found at the three observer positions. These proton signals are also observed in the X-band ENDOR spectrum given in Fig. 2A, and they disappear when the observer position is moved outside the EPR spectrum of MCR_{red2}. In addition, a number of weakly coupled protons from the macrocycle and possibly from axial ligands with hyperfine couplings ≤ 4 MHz are observed. An assignment of these smaller couplings to particular protons is not possible with the available (powder) data.

Discussion

The pulse EPR data show that MCR_{red2} is distinguished by two sets of nitrogen ligands with hyperfine couplings that differ by about a factor of two. In the ENDOR experiments, nitrogens with hyperfine coupling in the range 20.2–26.6 MHz are observed, whereas in the HYSCORE spectra a nitrogen(s) with a

Fig. 3 **A** Q-band nitrogen HYSCORE spectra measured at the observer positions *I*, *II*, and *III* (see Fig. 1); **B** simulation for field position *II*. The cross-peaks are labeled in the following way: *sq* (single-quantum) and *dq* (double-quantum) frequencies in the α or β electron spin manifolds; *superscripts* refer to the nitrogens with strong (*s*) and weak (*w*) hyperfine couplings

hyperfine coupling in the range 11.8–16.0 MHz is identified. The two vastly different nitrogen couplings indicate that the macrocycle is significantly distorted, a finding which is corroborated by the large rhombicity of the *g* matrix.

Many rhombic five- and six-coordinated nickel(I) and nickel(III) complexes are described in the literature [21, 22], but in most of them the smallest *g* value is very close to $g_e = 2.0023$, and the average *g* value, $g_{av} = (g_1 + g_2 + g_3)/3$, is much smaller than the one found for MCR_{red2}.

g values similar to those found in MCR_{red2} have been reported, for example, for some Ni(I)(STPP) complexes (STPP = 5,10,15,20-tetraphenyl-21-thiaporphyrin), where the nickel is coordinated to three nitrogens and one sulfur ligand of the macrocycle [23], for the pentacoordinated nickel complex [Ni(I)(DAPA)(SPh)₂]; DAPA = 2,6-bis[1-(phenylimino)ethyl]pyridine, with two thiolate ligands in the equatorial plane and three nitrogens of the DAPA ligand occupying the remaining coordination sites [24], and for Ni(III) tetraphenylcarboxyporphyrin, Ni(III)(CTPP), an inverted porphyrin where the nickel ion is coordinated to three nitrogens and one carbon in the porphyrin plane [25] (see Table 1).

In the case of Ni(I)(STPP) it has been found that the rhombicity of the EPR spectra is mainly caused by the axial ligands, rather than by a replacement of one of the pyrroles by a thiophene. The *g* matrix of the complex Ni(I)(STPP)(SO₂), for example, is nearly axial, whereas for Ni(I)(STPP)(2,4-Me₂py)₂, with two axial 2,4-lutidine ligands, the *g* values are very close to those of MCR_{red2}. A similar behavior is found for the Ni(III)(CTPP) complexes, where again, depending on the axial ligands, the *g* matrix reflects approximately tetragonal symmetry or can be highly rhombic. In all these compounds the rhombicity was attributed to a ground state described as a linear combination of $d_{x^2-y^2}$ and d_{z^2} orbitals.

The MCR_{red2} nitrogen with the smaller hyperfine interaction ($A_{iso} = 13.8$ MHz) observed in the HYSCORE experiments is tentatively assigned to the pyrrole nitrogen of ring A of the cofactor F430 in MCR_{red2} (see Fig. 5). Ring A is more flexible than the rings B, C, and D, which are π conjugated. In MCR_{red2} we therefore assume that pyrrole ring A is slightly bent out of the plane of the macrocycle. This, together with the influence of the axial ligands, would lower the symmetry and cause a strong mixing of the *d* orbitals of the ground state. That this ¹⁴N hyperfine coupling represents an interaction with the nitrogen of the glutamine is unlikely, since the glutamine is assumed to coordinate via its oxygen to the Ni ion and the nitrogen is remote (~0.4 nm from the Ni ion in MCR_{silent}). Consequently, the isotropic hyperfine coupling of the glutamine nitrogen is expected to be much

Table 2 ^{14}N hyperfine and nuclear quadrupole parameters of the pyrrole nitrogens of the “corphin” macrocycle of $\text{MCR}_{\text{red}2}$ in comparison with other MCR forms and Ni(I) complexes

Complex		A_1 (MHz)	A_2 (MHz)	A_3 (MHz)	A_{iso}^b (MHz)	$ e^2qQ/h $ (MHz)	η	Ref.
$\text{MCR}_{\text{red}2}$	$\text{N}_{\text{B,C,D}}^a (\times 2)^c$	26.6	23.2	24.0	24.6	2.0	0.00	This work
	$(\times 1)^c$	26.2	20.2	21.0	22.5	2.5	0.19	
$\text{MCR}_{\text{red}2}$	N_A^a	16.0	13.5	11.8	13.8	2.4	0.15	This work
$\text{MCR}_{\text{red}1c}$		27.7	28.7	28.6	28.3	—	—	[9]
$\text{MCR}_{\text{red}1}$	$\text{N}_{1,3}$	36.0	27.0	30.6	31.2	2.4	0.33	[20]
	$\text{N}_{2,4}$	31.5	22.5	25.0	26.3	3.6	0.67	
Ni(I)F430	$\text{N}_{1,3}$	31	22	26.5	26.5	3.7	0.19	[26]
	$\text{N}_{2,4}$	34	27	26.5	29.1	2.7	0.11	
$\text{MCR}_{\text{ox}1}$		24.4	28.8	28.6	27.3	—	—	[9]
$\text{MCR}_{\text{ox}1}$	N_1	31.0	22.0	24.0	25.7	3.2	0.25	[20]
$\text{MCR}_{\text{ox}1}$	N_2	31.5	24.5	21.0	25.7	2.6	0.08	
$\text{MCR}_{\text{ox}1}$	N_3	33.0	24.0	24.5	27.2	3.2	0.25	
$\text{MCR}_{\text{ox}1}$	N_4	33.5	26.5	26.3	28.8	2.6	0.08	
$\text{Ni(I)(STPP)(SO}_2\text{)}$		29.5	28.0	34.51	30.7	—	—	[23]

^aLabels A, B, C, D indicate the pyrrole nitrogens of the macrocycle (see Fig. 5). Estimated errors are $\Delta A_i = \pm 0.7$ MHz, $\Delta |e^2qQ/h| = \pm 0.5$ MHz, $\Delta \eta = \pm 0.2$

^bCalculated value, $A_{\text{iso}} = (A_1 + A_2 + A_3)/3$

^cThe two equivalent hyperfine couplings most likely come from the nitrogens in ring B and D since they are geometrically opposite (see Fig. 5), and the third hyperfine coupling would then be due to the nitrogen in ring C. In the simulations, both the **A** and **Q** tensors are rotated with respect to the **g** matrix. For the two equivalent

nitrogens ($A_{\text{iso}} = 24.6$ MHz) the Euler angles are $[\alpha, \beta, \gamma] = [45, 0, 0]^\circ$, and for the third ($A_{\text{iso}} = 22.5$ MHz) and weakest coupled nitrogen ($A_{\text{iso}} = 13.8$ MHz), $[\alpha, \beta, \gamma] = [135, 0, 0]^\circ$. For all four nitrogens the axes of the largest absolute nuclear quadrupole value point along **A**₁ and the smallest absolute value along **A**₃. The axes of the largest hyperfine couplings (**A**₁) are expected to point along the Ni-N bond; in this case the **g**₁ and **g**₂ axes approximately bisect the N'-Ni-N bond angles [28]

Fig. 4 Q-band proton Davies-ENDOR spectra at the observer positions indicated in Fig. 1. The *asterisks* label signals from protons with approximately isotropic hyperfine couplings. The spectra are centered around the ^1H Larmor frequency ($\nu_{\text{H}} \approx 48$ MHz)

smaller than $A_{\text{iso}} = 13.8$ MHz. It is worthwhile emphasizing that both ENDOR and HYSCORE spectra recorded at field positions where only $\text{MCR}_{\text{red}1}$ is present did not contain a weaker coupled nitrogen. These data show that there is a significant difference in the spin density distribution and macrocycle geometry between $\text{MCR}_{\text{red}1}$ and $\text{MCR}_{\text{red}2}$.

Fig. 5 Schematic drawing of the active site of ligation of methyl-coenzyme M reductase with (H)S-CoM coordinated via its sulfur, and glutamine α -147 coordinated via its oxygen. The pyrrole rings are labeled A, B, C, and D

Based on the simulations of the $\text{MCR}_{\text{red}2}$ nitrogen ENDOR and HYSCORE spectra and on chemical arguments, we assume that the larger nitrogen hyperfine couplings arise from the three directly coordinated

nitrogens of rings B, C, and D (see Table 2). The isotropic part of the hyperfine couplings ($A_{\text{iso}} = 24.6$ and 22.5 MHz) are smaller than the nitrogen hyperfine couplings found in MCR_{red1} ($A_{\text{iso}} \approx 28.5$ MHz) [9, 14], Ni(I)F430 ($A_{\text{iso}} \approx 27.8$ MHz) [26], and Ni(I)(STPP)(SO₂) ($A_{\text{iso}} = 30.7$ MHz) [23], reflecting the lower spin density on the macrocycle and the stronger d_{-2} admixture.

From previous studies on related systems it is expected that the axis of the largest nitrogen hyperfine coupling will point along the Ni-N bond [26, 28], which for all four pyrrole nitrogens is in the plane spanned by $\mathbf{g}_1$ and $\mathbf{g}_2$ (see Table 2). This orientation implies that $\mathbf{A}_3$ and $\mathbf{g}_3$ are normal to the macrocycle plane, consistent with expectations from the Ni hyperfine tensor which has its largest value along the $\mathbf{g}_3$ axis [9, 10]. The SOMO seems still to be predominantly $d_{x^2-y^2}$ in character, since the largest nitrogen hyperfine couplings are closer to the values found for MCR_{red1} and MCR_{ox1} [20, 27], as well as to Cu(II)TPP ($d_{x^2-y^2}$ with $A \approx 40$ MHz [28]), than to complexes with a d_{-2} ground state {e.g. CoTPP(py) with $A \approx 3$ MHz [29]}. A more quantitative description of the ground state admixture and coordination geometry is not attempted here since it requires a more detailed description of the EPR parameters, including nickel and sulfur (from HS-CoM) data.

The isotropic proton hyperfine couplings observed in the ENDOR spectra of MCR_{red2} are tentatively assigned to the methylene protons of coenzyme M, which is assumed to be axially coordinated to the nickel ion. Evidence in support of this assignment comes by comparing ^1H hyperfine couplings observed in MCR_{red1} and the model compound Ni(I)F430. These complexes have a dominant $d_{x^2-y^2}$ SOMO and no large and isotropic proton hyperfine couplings have been observed in the ENDOR spectra ($A \leq 4$ MHz). This is consistent with the fact that the EPR spectrum of the essentially square planar complex MCR_{red1} is not dependent on the presence of coenzyme M, and that no evidence for axially coordinated ligands has ever been found for solutions of F430 in the Ni(I) oxidation state. The ground state of MCR_{red2} has a more pronounced d_{-2} character, which decreases the spin density on the macrocycle (and the isotropic part of the macrocycle ^1H hyperfine couplings) relative to those of the complexes MCR_{red1} and Ni(I)F430 with a dominant $d_{x^2-y^2}$ ground state (see nitrogen couplings in Table 2). It is thus unlikely that the large and nearly isotropic ^1H couplings of $A_{\text{iso}} = 6.7$ MHz and $A_{\text{iso}} = 10.4$ MHz observed in MCR_{red2} are due to protons of the macrocycle. Hence, the C_β protons adjacent to the spin-attracting sulfur of HS-CoM are candidates for the isotropic hyperfine couplings. These hyperfine couplings can probably be traced back to spin-polarization caused by spin density on the sulfur, with the couplings depending on the dihedral angles between the π orbital of the S and the CH bond [30]. Similar isotropic hyperfine couplings have also been observed in type II blue copper proteins for cysteine β -protons [31] and in iron-sulfur clusters [32, 33].

To verify this hypothesis and to prove that HS-CoM is indeed an axial ligand of MCR_{red2}, samples containing HS-CoM with a deuterated methylene group and/or ^{33}S -labeled sulfur are in preparation¹. These data, together with a more detailed analysis of the magnetic parameters of MCR_{red2}, will be published elsewhere.

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¹A preliminary CW EPR spectrum of a ^{33}S -labeled MCR_{red2} sample shows a line broadening along $\mathbf{g}_3$ that is due to hyperfine splittings from ^{33}S

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