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# Methylation of the nucleobases in RNA oligonucleotides mediates duplex–hairpin conversion

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## ABSTRACT

We have systematically investigated the duplex to hairpin conversion of oligoribonucleotides under the aspect of nucleobase methylation. The first part of our study refers to the self-complementary sequence rCGCGAAUUCGCGA, which forms a stable Watson–Crick base paired duplex under various buffer conditions. It is shown that this sequence is forced to adopt a hairpin conformation if one of the central 6 nt is replaced by the corresponding methylated nucleotide, such as 1-methylguanosine  $N^2,N^2$ -dimethylguanosine,  $N^6,N^6$ -dimethyladenosine ( $m^6_2A$ ) or 3-methyluridine. On the other hand, the duplex structure is retained and even stabilized by replacement of a central nucleotide with  $N^2$ -methylguanosine ( $m^2G$ ) or  $N^4$ -methylcytidine. A borderline case is represented by  $N^6$ -methyladenosine ( $m^6A$ ). Although generally a duplex-preserving modification, our data indicate that  $m^6A$  in specific strand positions and at low strand concentrations is able to effectuate duplex–hairpin conversion. Our studies also include the ssu ribosomal helix 45 sequence motif, rGACCm<sup>2</sup>Gm<sup>6</sup><sub>2</sub>Am<sup>6</sup><sub>2</sub>AGGUC. In analogy, it is demonstrated that the tandem  $m^6_2A$  nucleobases of this oligoribonucleotide prevent duplex formation with complementary strands. Therefore, it can be concluded that nucleobase methylations at the Watson–Crick base pairing site provide the potential not only to modulate but to substantially affect RNA structure by formation of different secondary structure motifs.

## INTRODUCTION

Up to now more than 90 differently modified nucleosides have been identified in cellular RNA (1). Modifications of high structural complexity mainly occur in tRNA (2,3). Structurally simple modifications, as represented by pseudouridine, 2'-*O*-ribose or base methylated nucleosides, are also found in rRNA,

mRNA and snRNA (4). The function of the majority of nucleoside modifications is far from being fully understood. In recent years evidence has accumulated that important cellular processes are dependent on the presence of modified nucleosides. These processes are mRNA and rRNA maturation, ribosome assembly, rRNA processing, translation of the genetic code and recognition of tRNAs (5–7).

With respect to RNA secondary and tertiary structure, modified nucleotides are generally understood to modulate the physico-chemical properties of an existing RNA fold (8–13). In our opinion, this strongly underestimates the structure-determining potential of modified nucleotides. Here we report that nucleosides that are methylated at the Watson–Crick base pairing site are capable of substantially affecting RNA structure by mediating folding into different secondary structure motifs. This is demonstrated by a comparison of selected sequences that exist in duplex conformations but change into hairpin conformations if single nucleobases are replaced by the corresponding methylated ones.

## MATERIALS AND METHODS

### RNA synthesis and purification

5'-*O*-DMT-2'-*O*-TOM-protected nucleoside cyanoethylphosphoramidites (A, C, G and U) were obtained from Xeragon AG (Switzerland).

Based on known procedures (14–21), we synthesized the methylated nucleosides 1-methylguanosine ( $m^1G$ ),  $N^2$ -methylguanosine ( $m^2G$ ),  $N^2,N^2$ -dimethylguanosine ( $m^2_2G$ ),  $N^6$ -methyladenosine ( $m^6A$ ),  $N^6,N^6$ -dimethyladenosine ( $m^6_2A$ ),  $N^4$ -methylcytidine ( $m^4C$ ), 3-methyluridine ( $m^3U$ ) and 1-methylinosine ( $m^1I$ ) and transformed them into 5'-*O*-DMT-2'-*O*-TOM-protected nucleoside phosphoramidites (22,23); we will report on that in detail elsewhere.

All oligoribonucleotides were synthesized on CPG supports on a Pharmacia Gene Assembler Plus following slightly modified DNA standard methods: detritylation, dichloroacetic acid/1,2-dichloroethane (4:96, 2 min); coupling, phosphoramidites/acetonitrile (0.1 M, 120  $\mu$ l, 2.5 min) were activated with benzyl thiotetrazole/acetonitrile (0.35 M, 360  $\mu$ l); capping, (i) Ac<sub>2</sub>O/*sym*-collidine/THF (10:10:80), (ii) *N*-methylimidazole/THF (16:84); oxidation, I<sub>2</sub>/H<sub>2</sub>O/pyridine/THF

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(3:2:20:75, 45 s). Amidite solutions, tetrazole solutions and acetonitrile were dried over activated molecular sieves overnight. All sequences were synthesized trityl-off.

Deprotection and cleavage of oligonucleotides from the solid support were achieved with MeNH<sub>2</sub> in EtOH (8 M, 0.75 ml) and MeNH<sub>2</sub> in water (40%, 0.75 ml) for 3–6 h at 33°C. The solution was then evaporated to dryness. Removal of the 2'-*O*-silyl ethers was performed by treatment with tetrabutylammonium fluoride trihydrate (TBAF·3H<sub>2</sub>O) in THF (1 M, 0.90 ml) for at least 12 h at room temperature. The reaction was quenched by addition of Tris-HCl (1 M, pH 7.4, 0.95 ml). The volume of the solution was reduced to 1 ml and directly applied to a Sephadex G 10 column (30 × 1.5 cm) controlled by UV detection at 260 nm. The product was eluted with water and evaporated to dryness.

All oligoribonucleotides were purified by ion exchange chromatography on a semi-preparative Dionex DNAPac column at 80°C. Flow rate 2 ml/min; eluant A, 25 mM Tris-HCl, 6 M urea, in H<sub>2</sub>O, pH 8.0; eluant B, 25 mM Tris-HCl, 0.5 M NaOCl<sub>4</sub>, 6 M urea, in H<sub>2</sub>O, pH 8.0; detection at 265 nm; gradient I (to check the purity of the crude products after deprotection), 0–60% B in A over 45 min; gradient II (for purification), Δ20% B in A over 30 min. Fractions containing the purified oligonucleotide were desalted by loading onto a C18 SepPak cartridge (Waters/Millipore), followed by elution with 0.1–0.2 M (Et<sub>3</sub>NH)HCO<sub>3</sub>, water and then H<sub>2</sub>O/CH<sub>3</sub>CN (6:4). Combined fractions containing the oligonucleotide were lyophilized to dryness.

The expected masses were confirmed for all oligoribonucleotides by MALDI-TOF mass spectrometry.

### Thermal denaturation studies

Absorbance versus temperature profiles were recorded at 250, 260, 265 and 270 nm on a Cary-1 spectrophotometer equipped with a multiple cell holder and a Peltier temperature control device. Each sequence was measured at five or six different concentrations ranging from ~1 to 100 μM. Sequences 1–4 were measured in buffer solutions of 10 mM Na<sub>2</sub>HPO<sub>4</sub>, pH 7.0, containing either 150 mM or 1.0 M NaCl. All other sequences 5–21 were measured at a salt concentration of 150 mM NaCl. Data were collected after a complete cooling and heating cycle at a rate of 0.7°C/min. Melting transitions were reversible for all sequences and essentially the same with respect to the four different wavelengths.

For sample preparation oligonucleotides were lyophilized to dryness, dissolved in the corresponding buffer from stock solutions and subsequently degassed. A layer of silicon oil was placed on the surface of the solution.

Δ*H*<sup>o</sup> and of Δ*S*<sup>o</sup> values for monomolecular melting transitions were derived from a two-state van't Hoff analysis by fitting the shape of the individual α versus temperature curve; values of Δ*H*<sup>o</sup> and of Δ*S*<sup>o</sup> for bimolecular melting transitions were derived from 1/*T* versus ln*c* plots according to Markey and Breslauer (24) and Turner and co-workers (25,26). Errors for Δ*H* and Δ*S* arising from non-infinite cooperativity of two-state transitions and from the assumption of a temperature-independent enthalpy are typically 10–15%. An additional error is introduced when free energies are extrapolated far from the melting transitions; errors for Δ*G*<sup>o</sup> are typically 3–5%.

### NMR spectroscopy

<sup>1</sup>H NMR NH spectra were acquired on a Bruker Avance 500 MHz DRX using excitation sculpting for suppression of the water signal (27) or, alternatively, on a Varian Unity 500, applying a selective excitation refocusing sequence employing selective pulses shaped according to the G4 [excitation (28); 2.62 ms, rf amplitude 1.74 kHz] or REBURP [refocusing (29); 1.4 ms, rf amplitude 4.47 kHz] profile, respectively. Both shaped pulses were centered at 13 p.p.m.

Each of the sequences rCGCGAAUUCGCGA **3**, rCGCm<sup>2</sup>GAAUUCGCGA **5** and rCGCm<sup>2</sup><sub>2</sub>GAAUUCGCGA **6** was measured at two different concentrations (0.30–0.40 mM and 0.07–0.10 mM) and at three different temperatures (40, 27 and 1°C) in 25 and 6 mM sodium arsenate buffer, pH 7.4, (Sigma) without further addition of sodium chloride. rCGCm<sup>1</sup>GAAUUCGCGA **4** was measured at concentrations of 2.2, 0.7 and 0.1 mM in 25, 12.5 and 6 mM arsenate buffer. rCGCGm<sup>6</sup><sub>2</sub>AAUUCGCGA **7**, rCGCGAm<sup>6</sup><sub>2</sub>AUUCGCGA **8**, rCGCGm<sup>6</sup>AAUUCGCGA **9**, rCGCGAAm<sup>3</sup>UUCGCGA **11** and rCGCGAAUm<sup>3</sup>UCGCGA **12** were measured at concentrations of ~0.2 mM at 26°C.

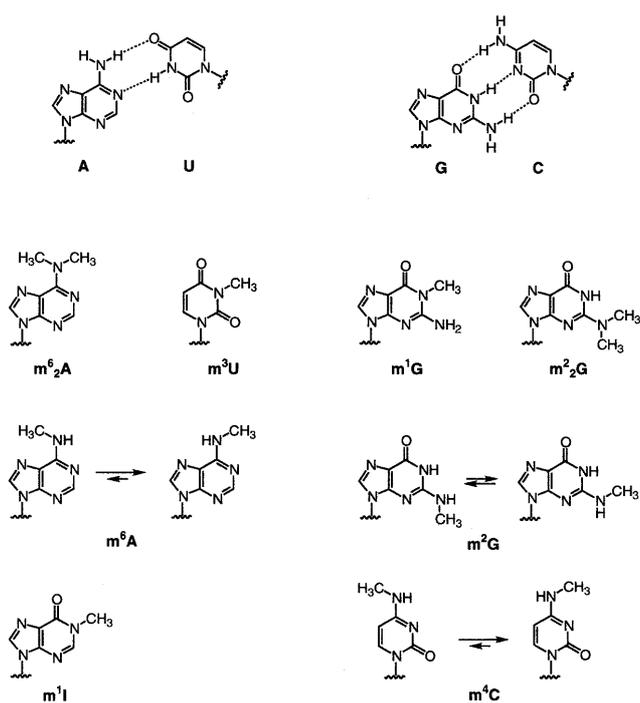
For sample preparation the oligoribonucleotides (triethyl ammonium salts) were lyophilized together with the corresponding amount of arsenate buffer, redissolved in water, lyophilized again and finally dissolved in H<sub>2</sub>O/D<sub>2</sub>O (9:1). The samples were equilibrated at room temperature; no heating and rapid cooling was performed.

### Electrophoretic mobility shift assay

PAGE was performed using gels prepared from solutions of 19% acrylamide, 0.8% bis(acrylamide), 100 mM NaCl, 2 mM MgCl<sub>2</sub>, 100 mM Tris-HCl, 0.12% tetramethylethylenediamine and 0.06% ammonium persulfate, pH 7.4 (15 × 15 × 0.1 cm). For sample preparation 7 μl of loading buffer (100 mM NaCl, 2 mM MgCl<sub>2</sub>, 100 mM Tris-HCl and 6% glycerine) were added to the lyophilized oligonucleotides **1–6**, **9**, **10**, **13** and **14** to obtain a final concentration of 45 μM. The samples were equilibrated at room temperature for 60 min. Electrophoresis was performed at room temperature. The electrophoresis buffer was 100 mM NaCl, 2 mM MgCl<sub>2</sub>, 100 mM Tris-HCl, pH 7.4. The gel was run at a constant voltage of 80 V, until the bromophenol blue marker had migrated ~5 cm. The gels were visualized using Stains All dye in formamide.

## RESULTS AND DISCUSSION

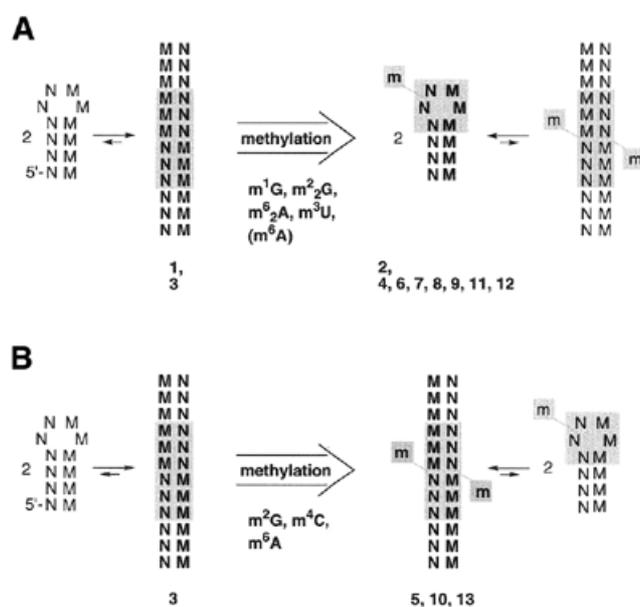
Among the naturally occurring methylated nucleosides we have focused on those that have a direct impact on the Watson-Crick base pairing pattern. The methyl groups of the nucleobases depicted in Figure 1 are located at sites that usually involve hydrogen bonding within a regular A-form duplex. Only in the cases of m<sup>2</sup>G, m<sup>6</sup>A and m<sup>4</sup>C rotamers exist that are capable of base pairing according to the Watson-Crick mode. For all other nucleobases shown (Fig. 1), namely m<sup>1</sup>G, m<sup>2</sup>G, m<sup>6</sup>A and m<sup>3</sup>U, a severe disturbance of the overall duplex structure would be expected if these nucleotides replaced the corresponding unmethylated nucleotide involved in a Watson-Crick base pair. Therefore, we asked whether a base pair disruption by methylation may also cause a change in secondary structure, provided the sequence itself allows this.



**Figure 1.** Watson-Crick base pairing pattern of adenine-uracil (A-U) and guanine-cytosine (G-C) (above). Chemical constitution and conformation of methylated nucleotides encountered in cellular RNAs: 1-methylguanosine ( $m^1G$ ),  $N^2,N^2$ -dimethylguanosine ( $m^2G$ ),  $N^6,N^6$ -dimethyladenosine ( $m^6A$ ), 3-methyluridine ( $m^3U$ ), 1-methylinosine ( $m^1I$ ),  $N^2$ -methylguanosine ( $m^2G$ ),  $N^6$ -methyladenosine ( $m^6A$ ) and  $N^4$ -methylcytidine ( $m^4C$ ). Preferred conformations of  $m^4C$ ,  $m^6A$  and  $m^2G$  are according to Engel and von Hippel (43).

A straightforward approach to probe this involves duplexes of palindromic or self-complementary sequences. A self-complementary sequence system is experimentally advantageous over a non-self-complementary one with respect to our intended comparison of methylated versus unmethylated oligoribonucleotides (Scheme 1). In principle, both methylated and unmethylated self-complementary sequences should be able to form either a hairpin or a duplex. Simple criteria, such as dependence of the melting transition on strand concentration as well as the shape of the UV melting profiles and their hyperchromicity, should allow easy distinction between monomolecular and bimolecular interactions and, subsequently, determination of the thermodynamic pairing parameters. Furthermore,  $^1H$  NMR NH resonances aid in distinguishing hairpin from duplex structures. A combination of these tools was used in the study presented. Additionally, for selected sequences gel mobility experiments were performed.

In general, hairpin formation is an intrinsic property of all palindromic sequences. In the DNA series the coexistence of hairpin and duplex structures is documented for several sequences, dCGGAATTTCGCG being the most famous among them (30–36). Low strand and low salt concentrations favor the hairpin, while high strand and high salt concentrations favor the duplex. Moreover, by heating and subsequent rapid cooling the hairpin conformations are kinetically trapped. Interestingly, in the RNA series palindromic sequences that establish a hairpin–duplex equilibrium are



**Scheme 1.** Site-specific methylation of a palindromic duplex. M, N, nucleotides capable of complementary Watson-Crick base pairing; mN, mM, methylated nucleotides. Depending on the kind of methylation the duplex is either converted into a hairpin (A) or the duplex is retained (B). The bold numbers refer to the sequences investigated (compare Table 1).

reported only for those that involve at least two mismatches within the corresponding duplex (37–41). This reflects the fact that RNA oligomers in micro- and millimolar concentrations thermodynamically prefer bimolecular over monomolecular interactions if formation of a fully Watson-Crick base paired duplex is possible. For example, this was observed for the Dickerson-Drew dodecamer analog, rCGCGAAUUCGCG **1** (Table 1). In buffer solutions containing 0.15 or 1.0 M NaCl this oligoribonucleotide showed a monophasic sigmoid melting profiles at neutral pH. A distinct dependence of the melting temperature on concentration (2–80  $\mu$ M) provided evidence that this sequence existed as a bimolecular complex. Therefore, we considered this sequence type suitable for our comprehensive methylation study. As a first test we incorporated  $m^1G$  instead of guanosine at strand position 4. Indeed, rCGCm<sup>1</sup>GAAUUCGCG **2** did show a monophasic sigmoid melting profile and, importantly, the independence of the melting temperature on concentration (2–90  $\mu$ M) proved the existence of a monomolecular transition, which is only consistent with a hairpin structure. Moreover, the significantly lower hyperchromicity was in accord with the lower number of base pairs.

The same effect of methylation was found in a closely related sequence system having a 3'-dangling adenosine. rCGCGAAUUCGCGA **3** existed as a duplex, whereas rCGCm<sup>1</sup>GAAUUCGCGA **4** formed a hairpin structure in buffer solutions containing up to 1 M NaCl (Fig. 2). Surprisingly, and highly interestingly, in 150 mM NaCl buffer solutions a comparison of thermodynamic parameters of the unmethylated sequences rCGCGAAUUCGCG **1** and rCGCGAAUUCGCGA **3** showed that the latter sequence with a dangling nucleotide was more unstable. This is a rare example of a

**Table 1.** Selection of methylated oligoribonucleotides and the corresponding reference sequences (5  $\mu$ M total strand concentration, 150 mM NaCl, 10 mM Na<sub>2</sub>HPO<sub>4</sub>, pH 7.0); melting temperatures  $T_m$ , thermodynamic data of double helix formation and mass spectrometry data<sup>a,b</sup>

No.	Base sequence	conformation	$T_m$ [a,b] [°C]	Thermodynamic data [a,b]			MALDI-TOF-MS	
				$\Delta H^\circ$ [kcal.mol <sup>-1</sup> ]	$\Delta S^\circ$ [cal.mol <sup>-1</sup> .K <sup>-1</sup> ]	$\Delta G^\circ_{298}$ [kcal.mol <sup>-1</sup> ]	[M + H] <sup>+</sup> (calc.)	[M + H] <sup>+</sup> (obs.)
1	CGCGAAUUCGCG	duplex	<b>62.2</b> (69.9)	<b>-88.1</b> (-109.0)	-138.5 (-293.5)	-17.0 (-21.5)	3811	3810
2	CGCm <sup>1</sup> GAAUUCGCG	hairpin	<b>54.2</b> (56.7)	<b>-35.3</b> (-34.2)	-107.2 (-102.6)	-3.3 (-3.6)	3825	3827
3	CGCGAAUUCGCGA	duplex	<b>58.5</b> (73.7)	<b>-49.8</b> (-126.6)	-126.0 (-340.7)	-12.2 (-25.0)	4141	4140
4	CGCm <sup>1</sup> GAAUUCGCGA	hairpin	<b>62.4</b> (65.0)	<b>-42.2</b> (-43.4)	-126.0 (-129.0)	-4.5 (-4.9)	4155	4153
5	CGCm <sup>2</sup> GAAUUCGCGA	duplex	<b>64.6</b>	<b>-100.7</b>	-273.8	-19.1	4155	4156
6	CGCm <sup>2</sup> <sub>2</sub> GAAUUCGCGA	hairpin	<b>64.8</b>	<b>-48.4</b>	-142.5	-5.9	4169	4168
7	CGCGm <sup>6</sup> <sub>2</sub> AAUUCGCGA	hairpin	<b>70.1</b>	<b>-46.9</b>	-136.6	-6.2	4169	4168
8	CGCGAm <sup>6</sup> <sub>2</sub> AAUUCGCGA	hairpin	<b>65.7</b>	<b>-48.0</b>	-140.3	-6.2	4169	4169
9	CGCGm <sup>6</sup> AAUUCGCGA	hairpin (< 5 $\mu$ M) duplex (> 40 $\mu$ M) <sup>c</sup>	<b>69.5</b>	<b>-31.0</b> <b>-113.8<sup>c</sup></b>	-92.5 -315.4 <sup>c</sup>	-3.4 -19.8	4155	4154
10	CGCGAm <sup>6</sup> AAUUCGCGA	duplex	<b>61.5</b>	<b>-103.7</b>	-284.4	-18.9	4155	4154
11	CGCGAAm <sup>3</sup> UUCGCGA	hairpin	<b>67.9</b>	<b>-48.5</b>	-141.9	-6.2	4155	4153
12	CGCGAAUm <sup>3</sup> UUCGCGA	hairpin	<b>68.0</b>	<b>-45.2</b>	-132.4	-5.7	4155	4156
13	CGCGAAUUm <sup>4</sup> CGCGA	duplex	<b>62.4</b>	<b>-107.1</b>	-293.7	-19.5	4155	4156
14	CGCm <sup>1</sup> IAAUUCGCGA	hairpin	<b>58.1</b>	<b>-39.3</b>	-117.7	-4.2	4140	4139
15	CGCGm <sup>1</sup> IAUUCGCGA	hairpin	<b>70.3</b>	<b>-47.6</b>	-139.6	-6.0	4156	4154
16	CGCGAm <sup>1</sup> IUUCGCGA	hairpin	<b>69.4</b>	<b>-49.7</b>	-145.7	-6.3	4156	4154
17	GACCGm <sup>2</sup> Gm <sup>6</sup> <sub>2</sub> Am <sup>6</sup> <sub>2</sub> AGGUC	hairpin	<b>64.6</b>	<b>-39.5</b>	-116.6	-4.7	3945	3945
18	GACCGGAAGGUC	hairpin	<b>72.4</b>	<b>-51.3</b>	-149.3	-7.1	3874	3873
19	GACCUUCCGGUC	hairpin	<b>64.2</b>	<b>-45.4</b>	-134.3	-5.4	3748	3748
20	GACCGGAAGGUC / GACCUUCCGGUC	duplex	<b>69.9</b>	<b>-129.7</b>	-353.7	-24.2		
21	GACCGm <sup>2</sup> Gm <sup>6</sup> <sub>2</sub> Am <sup>6</sup> <sub>2</sub> AGGUC / GACCUUCCGGUC	hairpins (< 5 $\mu$ M); competitive duplex formation at strand concentrations higher than 20 $\mu$ M						

<sup>a</sup>Terms in parentheses display the thermodynamic data for buffer conditions of 1.0 M NaCl, 0.01 M Na<sub>2</sub>HPO<sub>4</sub>, pH 7.0.<sup>b</sup> $\Delta H^\circ$  and  $\Delta S^\circ$  were determined from  $\alpha$  versus  $T$  plots by curve fitting (hairpins) and from  $1/T$  versus  $\ln c$  plots (duplexes) (24–26).<sup>c</sup>Thermodynamic data for the duplex were derived from a  $1/T$  versus  $\ln c$  plot of the four concentrations 36, 88, 149 and 313  $\mu$ M.

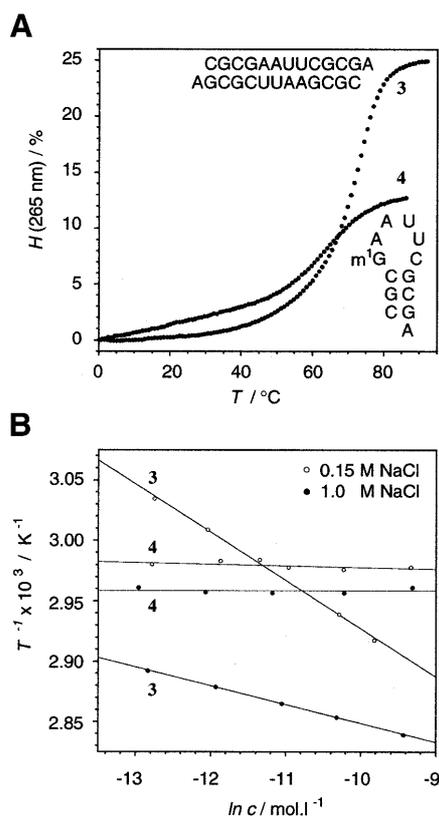
decrease in thermodynamic stability on attachment of a dangling nucleotide at the 3'-end of an oligoribonucleotide. Usually the attachment of a 3'-dangling nucleotide results in a profound increase in stability (25,26).

Although rCGCGAAUUCGCGA **3** did show a distinct dependence of melting temperature on concentration, we aimed at further proof of the exclusive occurrence of a duplex. In the <sup>1</sup>H NMR proton spectra of rCGCGAAUUCGCGA **3** at a total strand concentration of 70  $\mu$ M and in only 6 mM buffer solution without further addition of NaCl, six NH resonances were present, as expected for a symmetrical duplex structure of

12 bp (compare Fig. 3). One of the signals vanished at higher temperatures and was therefore assigned to the terminal base pair. Assignment of the remaining five NH protons was based on literature data for rCGCGAAUUCGCG (42).

The gel mobility data collected for sequences **1–4** were in accord with the observations described above (see Supplementary Material).

We considered the duplex of rCGCGAAUUCGCGA **3** at 150 mM NaCl as the most suitable sequence system for a comprehensive methylation study. On the one hand, with m<sup>1</sup>G we had a first example of a duplex–hairpin conversion. This

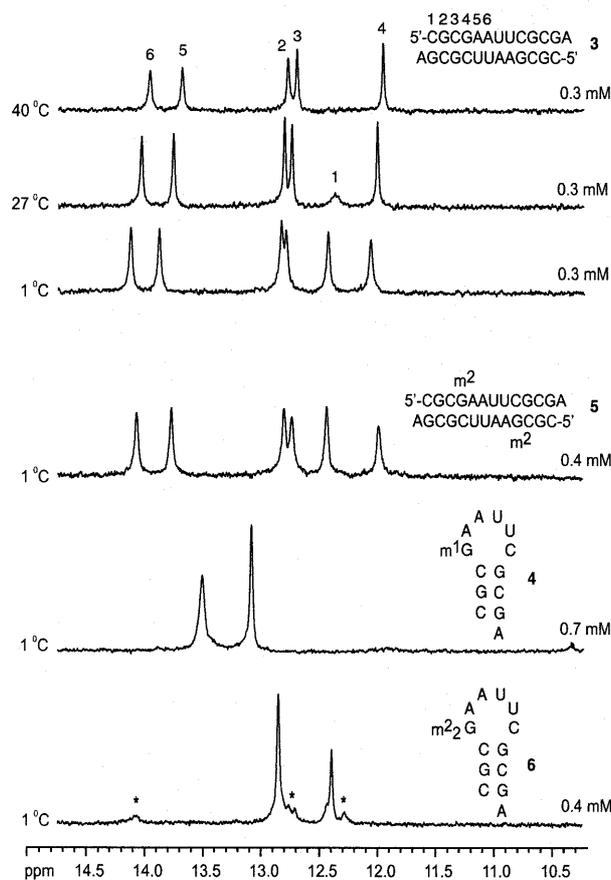


**Figure 2.** (A) Melting profiles of rCGCGAAUUCGCGA **3** and rCGCm<sup>1</sup>GAAUUCGCGA **4** (3  $\mu$ M, 1.0 M NaCl, 10 mM Na<sub>2</sub>HPO<sub>4</sub>, pH 7.0). (B) Dependence of melting transitions on strand concentration  $c$  of sequences **3** and **4**, (0.15 M or 1.0 M NaCl, 10 mM Na<sub>2</sub>HPO<sub>4</sub>, pH 7.0):  $1/T$  versus  $\ln c$  plots.  $H$ , hyperchromicity;  $T$ , temperature.

conversion proceeded in up to 1 M NaCl concentrations. It was therefore not a specific property of a duplex with a small enthalpy value as observed for this sequence at 150 mM NaCl concentrations (Table 1). On the other hand, a duplex providing such a small enthalpy value enabled us to discover which methylated nucleosides cause significant stabilization of the duplex itself. This will be outlined in the following paragraph.

#### Methylated nucleotides that interfere with regular interbase Watson–Crick hydrogen bonding for conformational reasons (m<sup>6</sup>A, m<sup>2</sup>G and m<sup>4</sup>C)

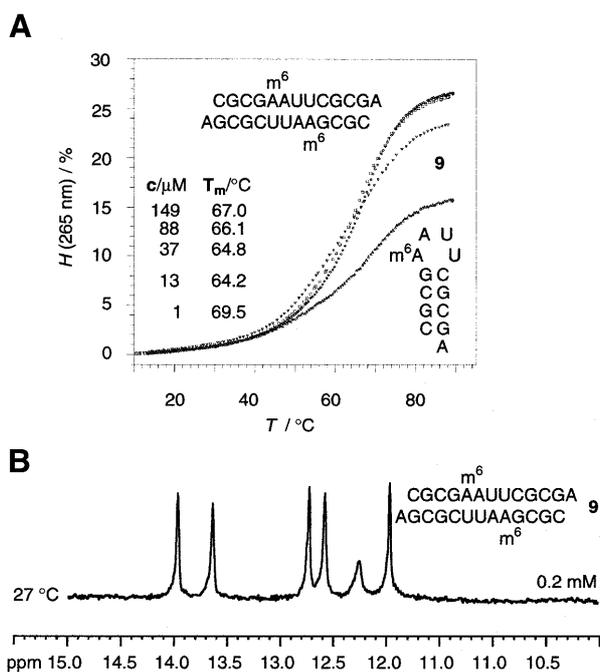
The chemical constitution of the monomethylated nucleotides m<sup>2</sup>G, m<sup>6</sup>A and m<sup>4</sup>C allows the formation of a standard Watson–Crick base pair. However, the conformation of the methylated exocyclic amino functionality may interfere and has to be taken into account. In the case of m<sup>2</sup>G, studies at the monomer level demonstrated that rotation of the *N*<sup>2</sup>-methyl-amino moiety was not restricted at room temperature (43). Therefore, it can be assumed that formation of a Watson–Crick base pair is not significantly hindered. Indeed, when the guanosine at position 4 of rCGCGAAUUCGCGA was replaced by m<sup>2</sup>G we even found a significant stabilization by 7 kcal/mol of the corresponding duplex rCGCm<sup>2</sup>GAAUUCGCGA **5**. Likewise, replacement of the adenosine at position 6 by m<sup>6</sup>A and



**Figure 3.** Selected <sup>1</sup>H NMR (500 MHz) NH spectra: rCGCGAAUUCGCGA **3** (duplex, assignment according to 42), rCGCm<sup>2</sup>GAAUUCGCGA **5** (duplex), rCGCm<sup>1</sup>GAAUUCGCGA **4** (hairpin) and rCGCm<sup>2</sup>GAAUUCGCGA **6** (hairpin, asterisks indicate signals for competitive duplex formation to a minor extent). Total strand concentrations are as indicated. H<sub>2</sub>O/D<sub>2</sub>O, 9:1; sodium arsenate buffer, pH 7.4; 25 mM for **3**, **5** and **6**, 12.5 mM for **4**.

replacement of the cytosine at position 9 by m<sup>4</sup>C resulted in a comparable stabilization of rCGCGAm<sup>6</sup>AUUCGCGA **10** and rCGCGAAUUm<sup>4</sup>CGCGA **13**. This duplex stabilization by ribonucleotides monomethylated at their exocyclic amino functionality may well reflect a specific property of the rCGCGAAUUCGCGA system in 150 mM NaCl buffer solutions, notwithstanding that Strobel and co-workers (44) recently ascertained m<sup>2</sup>G to be iso-energetic to guanosine in other selected self-complementary RNA duplexes and in a GNRA hairpin loop.

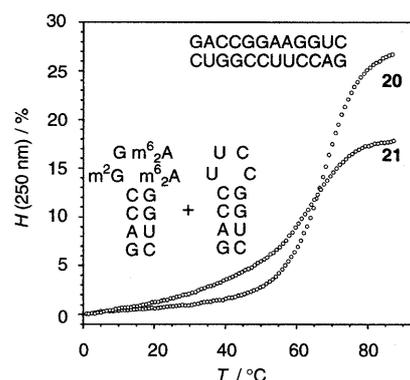
The higher duplex stability of rCGCGAm<sup>6</sup>AUUCGCGA **10** and rCGCGAAUUm<sup>4</sup>CGCGA **13** compared to that of rCGCGAAUUCGCGA **3** is not easily rationalized. From studies at the monomer level there is evidence that the methyl groups of m<sup>6</sup>A and m<sup>4</sup>C preferentially exist in the *syn* conformation (relative to N1 in the purine ring and relative to N3 in the pyrimidine ring) (43). Therefore, the formation of a Watson–Crick base pair involving m<sup>6</sup>A or m<sup>4</sup>C must be hindered to a small but definite extent and one is tempted to conclude that this hindrance would cause a decrease in pairing stability of the duplex. However, the preference for the *syn* conformation of the methyl group in m<sup>6</sup>A had an impact on the



**Figure 4.** (A) Examples of UV melting curves for the sequence rCGCG $m^6$ AAUUCGCGA **9**. Strand concentrations <5  $\mu$ M result in low hyperchromicities and invariable melting transitions of  $T_m$  = 69.5°C, indicating hairpin formation. Higher concentrations yield increased hyperchromicities and a linear  $1/T$  versus  $\ln c$  correlation (37–313  $\mu$ M,  $T_m$  = 64.2–69.2°C), indicating duplex formation. 150 mM NaCl, 10 mM Na<sub>2</sub>HPO<sub>4</sub>, pH 7.0. (B) <sup>1</sup>H NMR (500 MHz) NH spectrum of rCGCG $m^6$ AAUUCGCGA **9** at a concentration of 0.2 mM corroborates formation of a duplex (27°C, 25 mM arsenate buffer, pH 7.4).

duplex–hairpin equilibrium when this nucleotide replaced the adenosine at position 5. At 1 and 3  $\mu$ M concentrations rCGCG $m^6$ AAUUCGCGA **9** revealed monophasic melting profiles of low hyperchromicity with a melting transition at 69.5°C, indicating a monomolecular hairpin transition. At 13  $\mu$ M strand concentrations we observed a sigmoid melting profile of significantly higher hyperchromicity with a transition temperature at only 64°C. At higher concentrations (up to 313  $\mu$ M) the melting transition increased gradually and therefore reflected the occurrence of a bimolecular interaction, which is only compatible with a duplex structure. Accordingly, the six NH resonances found in the <sup>1</sup>H NMR spectrum measured for a strand concentration of 0.2 mM corroborate the existence of a duplex (Fig. 4). The gel mobility assay of sequence **9** indicated predominantly duplex formation for a loading concentration of 45  $\mu$ M in 100 mM NaCl, 2 mM MgCl<sub>2</sub> and 100 mM Tris–HCl buffer, pH 7.4 (see Supplementary Material).

However, the observation that at low strand concentrations (<5  $\mu$ M) monomethylated adenosine was able to initiate the duplex–hairpin conversion is remarkable. To the best of our knowledge, this structure-determining feature of  $m^6$ A has not been recognized so far and may enrich discussions about the reasons for frequent adenosine monomethylation in mRNA (45).



**Figure 5.** UV melting curves of rGACCGGAAGGUC/GACCUCCGGUC **20** and rGACCG $m^2$ G $m^6$ A $m^6$ AGGUC/GACCUCCGGUC **21**.  $c$  = 3  $\mu$ M. 150 mM NaCl, 10 mM Na<sub>2</sub>HPO<sub>4</sub>, pH 7.0.

### Methylated nucleotides that interfere with regular interbase Watson–Crick hydrogen bonding for constitutional reasons ( $m^2$ G, $m^6$ A, $m^1$ G and $m^3$ U)

The chemical constitution of the methylated nucleotides  $m^1$ G,  $m^2$ G,  $m^6$ A and  $m^3$ U does not allow formation of a standard Watson–Crick base pair. For all nucleotide replacements with a representative of this nucleotide class we observed a duplex–hairpin conversion of our reference sequence rCGCGAAUUCGCGA **3**. For instance, replacement of guanosine at position 4 by  $m^2$ G, as well as replacement of adenosine at either position 5 or 6 by  $m^6$ A resulted in the hairpin structures **6–8**. A similar effect was observed for replacement of guanosine at position 4 by  $m^1$ G and for replacement of uridine at either position 7 or 8 by  $m^3$ U, as represented by oligoribonucleotides **4**, **11** and **12**. The number of NH resonances displayed in the <sup>1</sup>H NMR spectra of **4**, **6–8**, **11** and **12** support the existence of hairpin structures (compare Fig. 3).

Two of the sequences mentioned in this paragraph, namely sequence **8** with  $m^6$ A at strand position 6 and sequence **11** with  $m^3$ U at strand position 7, showed a gradual appearance of a second melting transition at ~35°C for strand concentrations >20  $\mu$ M (see Supplementary Material). However, the <sup>1</sup>H NMR spectra did not indicate formation of a duplex for these sequences. Interestingly, formation of duplex to a minor extent was observed in the case of rCGC $m^2$ GAAUUCGCGA **6** (Fig. 3).

### Nucleotide replacement by 1-methylinosine ( $m^1$ I)

Nucleotide replacement with  $m^1$ I does not constitute a simple methylation of our reference sequence **3**. Nevertheless,  $m^1$ I was used in our study in order to shed light on the question of whether the hairpin rCGC $m^1$ GAAUUCGCGA is stabilized by a flipped base pair  $m^1$ G–C within the hairpin loop involving N3 and N2–H of the purine and N4–H and N3 of the pyrimidine (46). Replacing the guanosine at position 4 with  $m^1$ I closely resembled replacement with  $m^1$ G but, due to the missing exocyclic amino group, a bidentate flipped base pair cannot be formed. These considerations may explain the observed lower stability of the hairpin rCGC $m^1$ IAAUUCGCGA **14** (Table 1). A NMR investigation of rCGC $m^1$ GAAUUCGCGA is currently underway in the laboratory of B. Jaun in order to obtain supporting structural information.

$m^1I$  can also be regarded as a model for  $m^1A$ , a modification which is frequently found in cellular RNAs (4). Because of its reactivity and the easy hydrolyzation of the imino functionality,  $m^1A$  is not suited to incorporation into oligoribonucleotides by the phosphoramidite approach and to subsequent standard deprotection procedures. Replacement of the adenosines at positions 6 and 7 by  $m^1I$  did indeed result in hairpin structures, namely  $rCGCGm^1IAUUCGCGA$  **15** and  $rCGCGAm^1IUUCGCGA$  **16**. It is tempting to postulate that  $m^1A$  shows a similar behavior.

### Comparison with DNA

Detailed structural investigations on alkylation at the nucleobases of  $dCGCGAATTCGCG$  and closely related sequences, namely  $dCGCGAm^6ATTCGCG$  (47),  $dCGCm^6GAATTCGCG$  (48),  $dCGCm^6GAATTTTCGCG$  (49) and  $dCGCm^2IAATTCGCG$  (50), can be found in the literature. These sequences are observed as duplexes.

### Nucleobase methylation versus mismatches in RNA

RNA shows great tolerance for mismatches within a double helical arrangement. Numerous reports provide detailed thermodynamic and structural data on mismatches, such as G-U, G-A, A-A, G-G, C-A, C-U and others. Two examples relevant to our studies are given here. Morse and Draper (41) investigated the sequences  $rCGCGGCGCG$  and  $rCGCGACGCG$ . These sequences have the possibility to form hairpins but exist as duplexes with a central mismatch.  $rCGCGGCGC$  and  $rCGCGGCG$ , investigated by Kierzek *et al.* (51), also exist as duplexes.

Methylation of a nucleobase reduces the possibilities for hydrogen bonding and introduces a high steric demand. Both these features confine the number of energetically favorable mismatches between a methylated nucleobase and the former pairing partner. Consequently, the probability of retaining the duplex structure of a self-complementary sequence after methylation is decreased compared to after the introduction of a mismatch.

### Methylation of a non-self-complementary duplex: the ribosomal helix 45 motif $rGACCm^2GGm^6_2Am^6_2AGGUC$

The first part of our methylation study was based on a self-complementary sequence. Thereby, a single nucleotide replacement by a methylated counterpart inevitably generates a double interference in the corresponding duplex with methylated nucleobases located in opposite strands. This kind of methylation pattern is well known in DNA with respect to restriction-modification systems (52,53). In addition, we intended to expand our methylation studies to a non-self-complementary duplex.

Ribosomal helix 45 is the terminal helix at the 3'-end of the small subunit and is capped by a hypermethylated tetraloop, in which two successive  $m^6_2A$  residues represent rRNA modifications that have been conserved from bacteria to eukaryotes. Choosing a helix 45 sequence analog of *Bacillus stearothermophilus*,  $rGACCm^2GGm^6_2Am^6_2AGGUC$  **17**, which has been structurally characterized by Rife and Moore (54), we were interested in the questions whether methylation helps to preserve this hairpin fold and whether the unmethylated sequence is forced into a duplex in the presence of the complementary strand. This situation mimics a folding trap that is

avoided upon methylation. In this context, we also refer to recent computer simulations that demonstrate that RNA folding behavior is decisively improved by modifications (55,56).

In solution the unmethylated sequence  $rGACCGGAAGGUC$  **18** formed an extra-stable stem-loop structure of the GNRA type. However, upon addition of the complementary dodecamer  $rGACCUUCCGGUC$  **19** formation of a duplex (**20**) was unequivocally ascertainable from the perfectly sigmoid shape of the melting profile and the concentration-dependent melting transition.

We then investigated mixtures of the methylated counterpart  $rGACCm^2GGm^6_2Am^6_2AGGUC$  and  $rGACCUUCCGGUC$  **21**. The low hyperchromicity and a melting transition at 64°C, comparable to that of the individual strands, support the assumption of the coexistence of two individual hairpins, at least for concentrations <5  $\mu M$ . For concentrations >20  $\mu M$ , the gradual occurrence of a second transition at ~40°C indicated competitive duplex formation (Fig. 5).

Our current investigations focus on monomolecular helix 45 sequence constructs containing short complementary sequence partitions that are responsible for misfolds or two-state folding equilibria, as long as the sequences are unmethylated. These sequences fold into the correct hairpin structures upon methylation. We will report on that elsewhere.

### CONCLUSIONS

In the present study we have investigated the effect of methylation on the self-complementary oligoribonucleotide duplex of  $rCGCGAAUUCGCGA$ . Site-specific single nucleotide replacements by the corresponding methylated counterparts, such as  $m^2_2G$ ,  $m^6_2A$ ,  $m^1G$  and  $m^3U$ , caused conversion of our reference duplex into a hairpin structure. In contrast, the monomethylated nucleotides  $m^2G$  and  $m^4C$  behaved as duplex-preserving modifications and even yielded higher duplex stabilities. Remarkably, the monomethylated nucleotide  $m^6A$  demonstrated ambivalence with respect to duplex-hairpin conversion. At low strand concentrations our data indicated formation of a hairpin for the sequence  $rCGCGm^6AAUUCGCGA$ , whereas at higher concentrations formation of a duplex was ascertained. Furthermore, the effects arising from the methylated ribonucleotides were not limited to a self-complementary sequence context. In analogy, we obtained experimental evidence that tandem methylation of two successive adenosine nucleotides in the ribosomal helix 45 motif,  $rGACCm^2GGm^6_2Am^6_2AGGUC$ , prevents this hairpin structure from refolding into a duplex in the presence of the complementary sequence.

This basic study on RNA oligonucleotide methylation documents the potential of base-methylated nucleotides as primary determinants of RNA secondary structure.

### SUPPLEMENTARY MATERIAL

Supplementary Material is available at NAR Online.

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