Time-resolved X-ray absorption spectroscopy (TR-XAS) has so far practically been limited to large-scale facilities, to sub-picosecond temporal resolution and to the condensed phase. Here, we report the realization of TR-XAS with a temporal resolution in the low femtosecond range by developing a table-top high-harmonic source reaching up to 350 eV, thus partially covering the spectral region of 280 to 530 eV, where water is transmissive. We use this source to follow previously unexamined light-induced chemical reactions in the lowest electronic states of isolated CF$_4^+$ and SF$_6^+$ molecules in the gas phase. By probing element-specific core-to-valence transitions at the carbon K-edge or the sulfur L-edges, we characterize their reaction paths and observe the effect of symmetry breaking through the splitting of absorption bands and Rydberg-valence mixing induced by the geometry changes.
The application of X-ray sources to the study of the structure of matter has led to some of the most prominent advances in science in the 20th century, be it by diffraction (1) or spectroscopy (2). In the 21st century, the temporal dimension has been added to X-ray measurements, both at synchrotrons (3) and through the recent development of free-electron lasers (4–7). In parallel to these efforts, incoherent table-top hard X-ray sources have been applied to picosecond time-resolved studies (8,9). An alternative approach to generating soft X-rays with the advantages of full temporal and spatial coherence, as well as perfect temporal synchronization, is provided by high-harmonic generation. Very early efforts were successful at generating a modest soft-X-ray flux from titanium:sapphire drivers (10–12). Considerable progress came through the extension to longer-wavelength drivers (13–18). However, time-resolved measurements with such sources have so far remained out of reach.

Here, we describe a femtosecond time-resolved experiment using soft-X-ray supercontinua reaching into the water window. Using a long-wavelength driver (1.8 µm) with exceptional average power (2.5 W, i.e. 2.5 mJ at 1 kHz), we generate soft-X-ray supercontinua ranging from 100 to 350 eV, thus covering the chemically and biologically important K-edge of carbon. We use this source to study dissociation reactions of molecular cations, that have previously not been resolved in time, by transient absorption at the K-edge of carbon and all L-edges of sulfur simultaneously. This development considerably extends pioneering work on transient absorption in the extreme ultraviolet (19–22). These studies were limited to photon energies below ∼100 eV, and therefore to the L-edge of silicon and the M- and N-edges of heavier elements.

Our measurements probe the spatial structure of unoccupied orbitals and associated changes along the photochemical reaction pathways. Exploiting the sensitivity of X-ray absorption near-edge structure (XANES) spectroscopy to chemical shifts, we follow the evolution of the unoccupied valence orbitals of molecules from the neutral to the cation and along the reaction path to the final products. Using the sensitivity of dipole selection rules to molecular symmetry, we observe a splitting of some of the initially triply-degenerate orbitals of CF4 and SF6 to doubly- or non-degenerate orbitals as a consequence of the symmetry lowering induced by photodissociation.

In the experiment we focus a mid-infrared (MIR) femtosecond pulse centered at 1.8 µm into
Figure 1: **Soft-X-ray transient-absorption spectroscopy with a high-harmonic source** A) Experimental setup, B) High-harmonic spectrum at the carbon K-edge and transmitted spectrum through CF$_4$ gas, C) High-harmonic spectrum at the sulfur L-edges and transmitted spectrum through SF$_6$ gas.

A differentially-pumped high-pressure neon gas cell (Fig. 1A). A toroidal mirror images the soft X-ray (SXR) generation region into a target-gas cell where it is superimposed with a near-infrared (NIR) pulse centered at 800 nm. The SXR absorption of CF$_4$ (Fig. 1B) is dominated by overlapping transitions from the carbon 1s shell to the 5t$_2$ and 6t$_2$ unoccupied orbitals of dominant $\sigma^*$ character that peak at 298 eV. These assignments are based on our time-dependent density-functional theory (TDDFT) calculations of core-shell absorption spectra that use the LB94 functional and the QZ4P basis set (valence quadruple zeta + 4 polarization functions, relativistically optimized) and are consistent with high-resolution spectra recorded at synchrotron facilities (23). The SXR absorption spectrum of SF$_6$ (Fig. 1C) is dominated by transitions from the sulfur 2p$_{1/2}$ and 2p$_{3/2}$ shells to the a$_{1g}$ unoccupied orbital and to the shape resonances of t$_{2g}$ and e$_g$ symmetry lying above the L$_{2,3}$-edge. A weaker absorption feature at $\sim$ 240 eV is as-
signed to the transition from the sulfur 2s shell to the anti-bonding orbital of $t_{1u}$ symmetry, lying just below the $L_1$ edge, which is also consistent with the synchrotron literature (24). The weak modulation observed in both spectra up to energies of $\sim 180$ eV corresponds to harmonic structure. The experimentally determined resolution of $E/\Delta E = 308$ suggests that the appearance of these spectra is not spectrometer limited.

Figure 2: **Transient-absorption spectroscopy at the carbon K-edge** A) An intense near-infrared pulse induces single ionization of CF$_4$ to CF$_4^+$ which is unstable in its electronic ground state and dissociates into CF$_3^+$ + F. The sequence of geometries is taken from a calculated minimum-energy reaction path (see text for details). B) Absorbance ($A(t) = \ln(I_0/I(t))$) as a function of the SXR-NIR time delay. Negative time delays correspond to the SXR pulse preceding the NIR pulse. The intensity axis, as well as the color scale, are linear. The standard deviation of this data set amounts to 4%. The calculated stick spectrum in the inset has been shifted by -2.5 eV. C) Orbital diagram illustrating selected transitions, as obtained from TDDFT/LB94 calculations.

Strong-field ionization of neutral CF$_4$ in the gas phase is induced by an ultrashort NIR pulse centered at 800 nm, focused to a peak intensity of $(4 - 5) \times 10^{14}$ W/cm$^2$. Ionization of CF$_4$ to each of the three lowest-lying electronic states of CF$_4^+$, of respective vertical ionization energies 16.29, 17.51 and 18.54 eV (measured from the neutral ground state) (25), results in spontaneous
dissociation into CF$_3^+$ and F (26). At positive delays, an increase of the absorbance and a red
shift of its maximum are observed in the TR-XAS spectra (Fig. 2B). After reaching a maximal
absorbance, the absorption spectrum progressively splits into multiple bands, as illustrated by
the sample spectrum (averaged over delays of 300 to 500 fs) shown as inset in Fig. 2B. Most
prominently, one band is observed to shift to lower photon energies by 10 eV, terminating at 288
eV. Two further bands shift up and down by $\sim 1$ eV, respectively, and a fourth band appears as a
shoulder of the absorption spectrum around 302 eV. A 50% fraction of the static CF$_4$ absorption
spectrum has been subtracted from the whole data set (Fig. 2B) to account for partial ionization
of the probed sample. This fraction has been obtained by comparing the absorption spectrum
at long delays with calculated spectra of CF$_3^+$. The delay in the appearance of the isolated
absorption band of CF$_3^+$ has been determined by integrating the signal at each delay over a
narrow energy range and fitting an error function to the time-dependent signal. This analysis
reveals a time delay of $(40 \pm 2)$ fs in the appearance of the absorption band and provides a time
resolution of $(40 \pm 5)$ fs (SOM, Section 2.2).

The observed changes in the absorption spectrum are the signature of symmetry lowering
that occurs when the initially tetrahedral CF$_4^+$ molecule dissociates into the trigonal planar CF$_3^+$
molecule. Descent-in-symmetry arguments from group theory combined with dipole-selection
rules show that a transition of the type C 1s → $t_2$ in CF$_4$ must split into two transitions of the
type C 1s → $a''_2$ and C 1s → $e'$ in CF$_3^+$, as observed in Fig. 2B and illustrated in Fig. 2C.

Further insight is obtained by comparison with advanced quantum-chemical ab-initio calcula-
tions (stick spectrum in Fig. 2B and Fig. 3), which are further described in the supplementary
online material (SOM, Section 3). The X-ray absorption spectrum of CF$_4$ is dominated by
transitions to the 5$t_2$ and 6$t_2$ orbitals, which appear as two intense lines at the bottom of Fig.
3A. Along the reaction path, the transition to the 5$t_2$ orbital first splits into transitions to three
non-degenerate orbitals in CF$_3^+$. This is the signature of the Jahn-Teller effect which causes the
minimum-energy geometry to be of C$_{2v}$ symmetry for short ($\leq 1.6$ Å) C-F bond lengths (27).
With increasing bond length, the minimum-energy geometry changes to D$_{3h}$, which is accom-
panied by the merging of the second- and third-lowest transitions to a single line corresponding
to a transition to the 5$e'$ orbital, which confirms the symmetry arguments given in the preceding
paragraph.
Figure 3: **Calculated X-ray absorption spectra of the reaction CF$_4^+$ → CF$_3^+$ + F**

A) X-ray absorption spectra, calculated with the TDDFT method using the LB94 functional and the QZ4P basis set, as a function of one C-F internuclear separation along the minimum-energy reaction path CF$_4^+$ → CF$_3^+$ + F calculated on the CCSD/6-31G* level of theory. The CF$_4$ X-ray absorption spectrum is shown below the dotted line. A linear intensity scale has been used. B) Unoccupied orbitals characteristic of the final state of the X-ray transition, corresponding to the dominant absorption bands of panel A. Only one of three equivalent orbitals is shown in the tetrahedral (T$_d$) geometry of CF$_4^+$, whereas both, distinct orbitals of e’ symmetry are shown for CF$_3^+$.

Figure 3A also reveals a very large shift of the lowest-energy X-ray-absorption band occurring in the transition from CF$_4^+$ to CF$_3^+$. The major part of this shift originates from the changing energy of the unoccupied $2a''_2$ orbital, caused by its evolution along the reaction path as shown in Fig. 3B. We next consider the complementarity of the intensity evolutions in the 5e’ and 6e’ transitions. In the tetrahedral geometry of CF$_4$, the $5t_2$ orbital has dominant valence character, whereas the $6t_2$ orbital has dominant Rydberg character with most of its probability located outside the sphere defined by the fluorine atoms. This arises because the high electron density at the strongly electronegative fluorine atoms creates an effective potential barrier along the radial direction. This so-called ”cage effect” has first been discussed in the context of XANES spectra of SF$_6$ to which we return below (28). As the CF$_4^+$ molecule dissociates, the fluorine cage opens up, going from a tetrahedral geometry to a planar one in CF$_3^+$. This evolution causes
a mixing of Rydberg and valence character of the orbitals. Indeed, orbitals previously localized inside (valence) or outside (Rydberg) the cage display a mixed character after dissociation. This change causes a decrease of the overlap between the highly localized core orbital and the initially localized valence orbitals, translating into a decrease of their absorption as the cage opens. In contrast, Rydberg orbitals with initially small overlap with the core orbital exhibit an increase of their absorption as the cage opens up. Along the dissociation coordinate the 5e’ orbital therefore develops partial Rydberg character, whereas the 6e’ orbital acquires partial valence character (see Fig. 3B). This explains both the calculated and the observed intensity variations, i.e. a decreasing intensity for the transition leading to the 5e’ orbital and an increase of the transition strength to the 6e’ orbital. The appearance of the shoulder on the high-energy side of the absorption spectrum around 302 eV (Fig. 2B) is also reproduced by the calculations shown in Fig. 3A and is attributed to the 7t₂ orbital acquiring partial valence character as it evolves into the 7e’ orbital of CF₃⁺.

The good agreement between experimental and theoretical results enabled us to reconstruct the average C-F internuclear separation as a function of time during the dissociation of CF₄. These results are shown in Section 4.2 and Fig. S10 (SOM). Moreover, the analysis of time-dependent energy shifts of the 2a'' absorption band provides evidence for the transient excitation of vibrational modes during the dissociation process (Section 4.3 and Fig. S11 in SOM).

We proceeded to demonstrate the generality of our technique by turning to the sulfur L-edges and studying the photodissociation SF₆⁺ → SF₅⁺ + F. The three lowest electronic states of SF₆⁺ of respective vertical ionization energies 15.7, 17.0 and 17.0 eV (measured as energy difference from the neutral ground state) (29), all dissociate to SF₅⁺ + F (26).

Figures 4B and 4C show the observed time-resolved XANES spectra following strong-field ionization of SF₆, observed at the L₂,₃-edge and the L₁-edge of sulfur simultaneously. A 55% fraction of the static SF₆ absorption spectrum has been subtracted from the whole data sets (Figs. 4B and 4C) for clarity, for the same reason as in Fig. 2B. The L₂,₃-XANES spectrum at negative delays is dominated by transitions from the sulfur 2p shell of SF₆ to the unoccupied orbital of a₁g symmetry, lying below the L₂,₃ edge, and to the shape resonances of t₂g and e₉g symmetry lying above (see Fig. 4D). These shape resonances are confined by the effective potential barrier created by the presence of the surrounding electronegative fluorine atoms (28),
Figure 4: Transient absorption spectroscopy at the sulfur L-edges
A) An intense infrared pulse induces single ionization of SF$_6$ to SF$_6^+$ which is unstable in its electronic ground state and dissociates into SF$_5^+$ + F. The sequence of geometries is taken from a calculated minimum-energy reaction path. B) Absorbance ($A(t) = \ln(I_0/I(t))$) at the L$_{2,3}$-edge as a function of the SXR-NIR time delay. Negative time delays correspond to the SXR pulse preceding the NIR pulse. The standard deviation of this data set amounts to 4.3%. C) Absorbance at the L$_1$-edge of SF$_6$. The insets show absorption spectra obtained by averaging over delays of 200-400 fs. All intensity axes shown in this figure, as well as all color scales, are linear. The standard deviation of this data set amounts to 6.3%. D) Orbital diagram illustrating the experimentally observed transitions.

as we have discussed in the case of CF$_4$. However, in SF$_6$, they are contained within the cage and therefore have considerable overlap with the central sulfur atom, which explains their strength in the experimental absorption spectra. The $a_{1g}$ band shifts to lower energies by 2 eV, the $t_{2g}$ band
splits into two main components that shift to lower energies by 2.5 and 6.5 eV, respectively, and the $e_g$ band shifts to lower energies and broadens considerably. The red shift of all of these absorption bands is consistent with an increase of the local electron density on the sulfur atom, which is a consequence of the loss of an electron-withdrawing fluorine atom. The delay in the appearance of the absorption bands of SF$_5^+$ has been determined by integrating the signal at each delay over a narrow energy range and fitting an error function to the time-dependent signal. This analysis reveals time delays ranging from $(34 \pm 4)$ fs to $(74 \pm 10)$ fs in the appearance of the absorption bands and provides a time resolution of $(45 \pm 7)$ fs (SOM, Table S1 and Section 2.3).

The absorption at the L$_1$-edge (Fig. 4C) is dominated by the transition to a $t_{1u}$ anti-bonding orbital, which is observable because of the gerade parity of the 2s initial orbital. This band splits into two main components that shift to lower energies by 6 and 15 eV, respectively.

The changes observed in the TR-XAS spectra of SF$_6$ can again be assigned by group theory. SF$_6^+$ in each of its lowest three electronic states can only dissociate into SF$_5^+$ in its electronic ground state, which has a trigonal bipyramidal geometry and belongs to the D$_{3h}$ point group. By symmetry, the transition to the $a_{1g}$ orbital of SF$_6$ should correlate with a transition leading to an $a'$ orbital in the D$_{3h}$ point group of SF$_5^+$. Similarly, the $t_{2g}$ shape resonance of SF$_6$ will correlate with an $e'$ or $e''$ shape resonances in SF$_5^+$. Finally, the $e_g$-symmetry shape resonance should correlate with an $e'$- or $e''$-symmetric resonance in SF$_5^+$. These symmetry correlations precisely describe the experimental observations. The absorption bands corresponding to the unoccupied $a_{1g}$ orbital and the $e_g$ shape resonance are both observed to shift to lower energies, without apparent sign of splitting. In contrast, the absorption band corresponding to the $t_{2g}$ shape resonance is observed to split into two bands. Whereas the shift of all transitions happens almost simultaneously (SOM, Figs. S12 and S13), the splitting of the $t_{2g}$ band is delayed by $\sim 70$ fs. Calculations (SOM, Section 5) suggest that this delayed splitting is caused by a higher sensitivity of the valence-type orbital characteristic of the $t_{2g}$ shape resonance in SF$_6$ to bond angle changes, as compared to the $a_{1g}$ and $e_g$ orbitals (SOM, Fig. S16). The observed splitting of the absorption band would then be driven by rapid changes in bond angles of the SF$_5^+$ unit that occur around values of 2.5-3 Å for the SF$_5^+$-F dissociation coordinate.

Qualitatively similar dynamics are observed at the L$_1$ edge. By symmetry, the dominant $2s \rightarrow t_{1u}$ transition can only split into two transitions leading to orbitals of $e'$ or $a''_2$ symmetries,
which corresponds well to the experimental observation of a splitting into two bands. The broad absorption feature observed around 267 eV corresponds to several absorption bands, all located above the L$_1$ edge (244.17 eV). These bands correspond to shape resonances which have a very short lifetime and, therefore, a large natural linewidth leading to the observation of a single very broad absorption feature.

Our results demonstrate the feasibility of time-resolved XAS with table-top light sources and its potential in elucidating the dynamics of electrons and nuclei in chemical reactions. Specifically, this method nicely complements other key techniques in molecular reaction dynamics, such as time-resolved photoelectron spectroscopy (30) and time-resolved high-harmonic spectroscopy (31). We therefore anticipate TR-XAS to become a decisive technique for the investigation of non-adiabatic molecular dynamics, such as those occurring at conical intersections (32, 33). Owing to its sensitivity to elements, TR-XAS will enable time-resolved studies of electronic dynamics with atomic spatial sensitivity. Broadening the spectral coverage of our source only slightly would bring time-resolved extended X-ray absorption fine structure (TR-EXAFS) in combination with X-ray absorption near edge structure (XANES) experiments within reach, providing full structural and electronic information. A further improvement of the photon flux may bring non-linear X-ray spectroscopies within reach (34). Although demonstrated in the gas phase, our method is directly applicable to the solid state and is readily extendable to the liquid phase using flow cells (35) or flat microjets (36).

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Additional data supporting the conclusions are shown in the supplementary material.

References and Notes


